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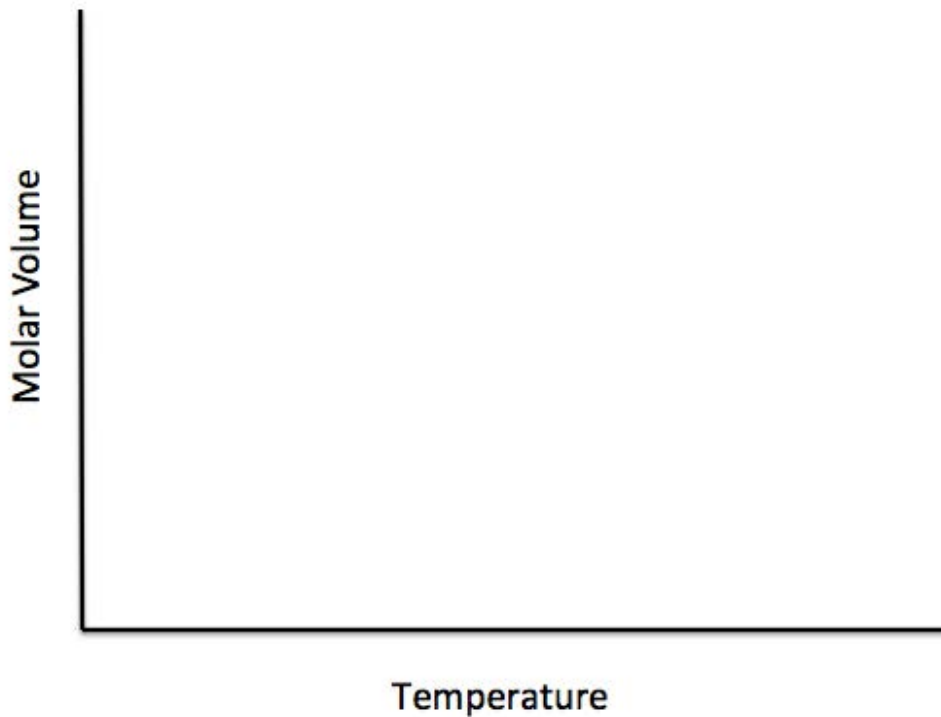
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**3.091 Introduction to Solid State Chemistry**  
**Fall Term 2018**  
**Quiz 8 (A)**

Do yourself a solid.

1) Shown below is a processing pathway for creating a glass from a crystal.

(a) Label the melting temperature(s) ( $T_m$ ) and the glass transition temperature(s) ( $T_g$ ) on the temperature axis: (1 point)



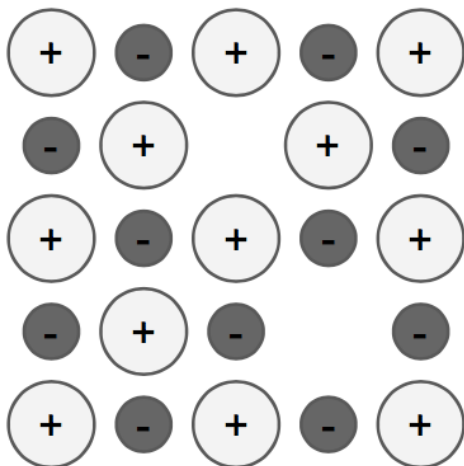
(b) Label the line segments I, II, III, and IV crystalline, amorphous, liquid, or super cooled. (2 point)

(c) Rank A, B, and C from least to greatest disorder. Explain in one sentence how you know. (1 point)

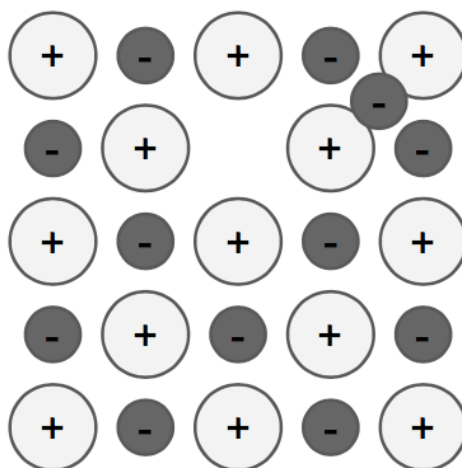
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(d) Is the cooling rate of A or B greater? (1 pt)

2) Fill in the blanks: (2.5 pts)



\_\_\_\_\_ Defect  
 is made of two \_\_\_\_\_



\_\_\_\_\_ Defect  
 is made of one \_\_\_\_\_  
 and one \_\_\_\_\_

3) Look at your atomic model. Isn't it pretty? You shake it vigorously and see that 12 vacancies have formed. This represented the material at  $T = 900\text{K}$ . You then gently tap the side of the model until only 7 vacancies remain, representing your material at  $T = 500\text{K}$ . Assuming 800 atom sites in your model, what is the vacancy formation energy for this material (in eV)? (2.5 pts)

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